

Unit-I (Atomic Structure)

I. ~~each~~ one marks: (5x1 = 5)

1. de-broglie equation
2. The hybridisation of IF_7 molecule
3. The bond order of Nitrogen molecule
4. Intra molecular H-bonding present in
5. ~~sub~~ Dual character is explained by

ii. 3 marks (5x3 = 15)

1. state Heisenberg's uncertainty principle.
2. Define Hybridisation.
3. Define bond order.
4. Define molecular orbitals.
5. Write significance of negative electron energy.

iii. 5 marks 6x5 = 30.

1. De-broglie equation.
2. Postulates of molecular orbital theory.
3. Energy level diagram of N_2 molecule Explain.
4. Energy level diagram of O_2 molecule Explain.
5. Davison gessmer Experiment Explain.
6. Importance of Hydrogen bonding.