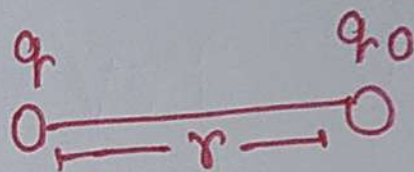


### (3) Electric field ( $\vec{E}$ )

(direction of  $E \Rightarrow$  '+' to '-')

$$\vec{E} = \frac{\vec{F}}{q_0}$$



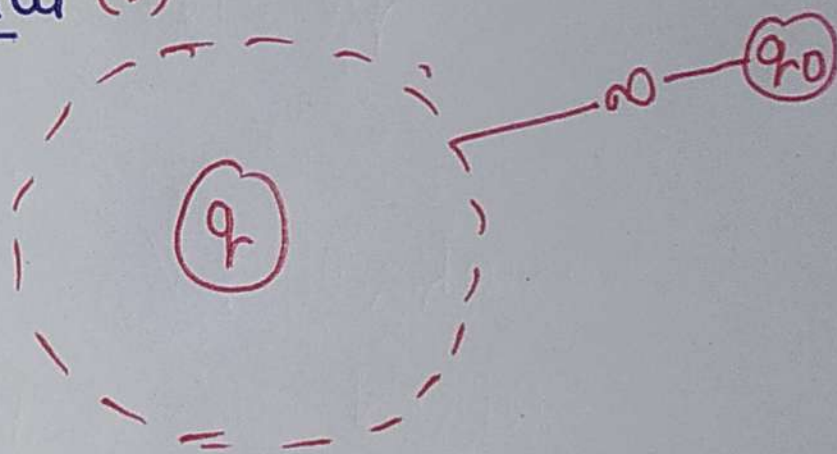
$$E = \frac{k q}{r^2}$$

unit of  $E = N/C$

### (4) Electric Potential ( $V$ )

$$V = \frac{W}{q_0}$$

$$V = \frac{k q q_0}{r}$$



$$W = \frac{k q q_0}{r} = V q_0$$

Relation b/w  $V$  and  $E$

$$V = E \cdot d$$

so,  $W = V q_0$

$$W = q_0 E d$$

unit of  $W = \text{Joule}$   
unit of  $V = \text{volt}$

# Electrostatic

\* Unit of Charge - Coulomb

$$1 \mu\text{C} = 1 \times 10^{-6} \text{C}$$

$$1 \text{ nC} = 1 \times 10^{-9} \text{C}$$

## (1) Quantization of Charge

$$q = ne$$

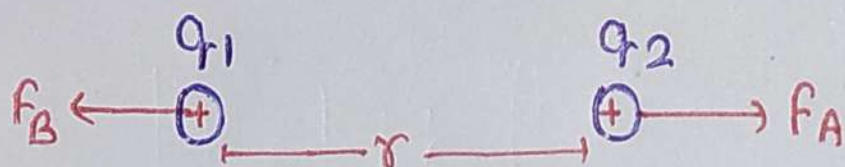
deficiency of  $e^-$  = '+' charge

excess of  $e^-$  = '-' charge

$$e = 1.6 \times 10^{-19} \text{C}$$

$n$  = no. of electrons

## (2) Coulomb's force (Electrostatic force)



Force on charge particle 2 due to 1 is  $F_A$

$$F_A = k \frac{q_1 q_2}{r^2} \quad \left( k = 9 \times 10^9 \frac{\text{N} \cdot \text{m}^2}{\text{C}^2} \right)$$

Force on charge particle 1 due to 2 is  $F_B$

$$F_B = k \frac{q_1 q_2}{r^2} \quad \left( \text{unit of } F = \text{Newton} \right)$$
$$= \frac{\text{kg} \cdot \text{m}}{\text{sec}^2}$$

So,

$$\boxed{F_A = F_B}$$

That means the ratio of force due to both charges is always  $1:1$ .