

atoms?

CBSE TEST PAPER-03

CLASS - IX Science (Atoms and Molecules)

1. Which of the following statements is correct? [1] (a) Cathode rays travel in straight line and have momentum. (b) Cathode rays travel in straight line and have no momentum (c) Cathode rays do not travel in straight line but have Momentum. (d) Cathode rays do not travel in straight line and have no momentum. β –particles are represented as :- (a) $_{-1}^{0}e$ (b) $_{+1}^{0}e$ (c) $_{-1}^{1}e$ (d) $_{0}^{1}e$ 2. [1] $^{40}_{18}Ar$ and $^{40}_{20}Ca$ are (a) Isotopes (b) Isobars (c) Isotones (d) Both b and c 3. [1] The maximum number of electrons in L shell is (a) 8 (b) 18 (c) 28 (d) 38. 4. [1] 5. What are α , β and γ – rays composed of. [2] What is meant by e/m ratio? What was the value of this ratio for a particle in the 6. [2] cathode rays? Complete the following equations which describe nuclear charge -7. [2] a) ${}_{11}^{23}Na + {}_{0}^{1}n \rightarrow \underline{\hspace{1cm}} + {}_{0}^{0}e$ b) ${}_{4}^{9}Be + {}_{2}^{4}He \rightarrow {}_{0}^{1}n +$ c) ${}_{8}^{16}O + {}_{0}^{1}n \rightarrow$ 8. Stat the properties of cathode rays? [2] 9. Write an experiment to show that cathode rays travel in straight line? [3] 10. What is radioactivity? What are the applications of radioisotopes? [3] 11. There are 2 elements C and B. C emits an α – particle and B emits a β – particle. [3] How will the resultant elements charge? 12. What are isotopes? Name the isotopes of hydrogen and draw the structure of their