1.	Define following terms				5×2=10
	(i) Evaporation		(ii) Boiling point		
	(iii) Sublimation		(iv) Colloidal solutio	n	(v) Suspension
2.	Write 4-Factors which effe	ect the evaporation			4
3.	Why evaporation causes co	ooling?			2
4.	Write 4-difference between colloidal and suspension				4
5.	5. What is compound and Heterogeneous mixture				
6.	6. Write electronic configuration of Following element				4
	(a) Na	(b) B	(c) Ca	(d) Ar	
7.	Which of following are mi	xture, compound?			5
	(i) salf solution	(ii) Sugar solution	(iii)	Alloys	
	(iv) milk	(v) Calsium carbonate	(vi) S	Sugar	
	(vii) blood	(viii) Carbon dioxide	(ix)	oxygen	
	(x) Soil				
8.	Difference Homomoseneo	2			
	<b>9.</b> Define following term with example?				[2×6=12]
	(a) compound	(b) Homogeneous	(c) colloidal		
	(d) Suspension	(e) Tyndal effect	(f) Brownian	movemen	ıt.
	10. Which of the following				
	[3]				
	Ca, Na, K, P, Si, C, N				
	11. Write No. of valence e	[3]			
	Mg, Al, Cl, Ar, Ca, B, N				
	<b>12.</b> Write atomic number a	[3]			
	Si, P, S, F, O, C				
	<b>13.</b> Define:-				
	$[2 \times 4 = 8]$				

	(i) Boiling point	(ii) melting point		
	(iii) Evapourisation	(iv) sublimation		
	ime.			
	[2]			
	<b>15.</b> Name four metalloids?		[2]	
16. What are inert gas? Why inert gas donot form compound.				
	[2+1=3]			
	17. Name five metale and five non-metal.		[4]	